## LAB-SCALE PRODUCTION OF RHAMNOLIPID BY *PSEUDOMONAS AERUGINOSA* A3 AND STUDY ITS SYNERGISTIC EFFECT WITH CERTAIN ANTIBIOTICS AGAINST SOME PATHOGENIC BACTERIA

A. M. Alshaikh Faqri N. H. Hayder A. J. Hashim Researcher Prof. Prof. Department of Biotechnology, College of Science, University of Baghdad Email: ayooboy@scbaghdad.com

#### ABSTRACT

This study was aimed to obtain the considerable amount of the biosurfactant (rhamnolipid) from local isolate of Pseudomonas aeruginosa on industrial-scale. A 7L laboratory bioreactor was appointed to accomplish this task and all conditions of production were set as previously by batch culture experiments. Bioreactor with 3.5 L working volume was conducted for 120 h to obtain 16 g/L at 30°C and 6.5 pH. The supernatant of production medium was used to reduce the surface tension of water from 72 to 27 mN/m, and the emulsification index was reached to 67%. The product was extracted with a mixture of solvents, purified by silica gel using a column of glass chromatography  $(3.5 \times 30 \text{ cm})$ , and characterized by TLC and FTIR where it turn out that the biosurfactant is composed of both mono- and di-rhamnolipid. Mono-RL was tested for its antibacterial activity against some clinical isolates of Grampositive and Gram-negative bacteria to obtain the MIC, which was about 250µg/ml, also at same concentration, the toxicity against Rat embryonic fibroblast (REF) cell line was reached to 23%. The combination effect of Mono-RL with some antibiotics was studied against pathogenic clinical isolates revealed that there is a synergistic effect with ampicillin against S. aureus and Serratia marcescens, with cefotaxime against E. coli and Kelbsiella pneumoniae, same as with tetracycline.

Keywords: biosurfactant, bioreactor, emuslification, chromatography

مجلة العلوم الزراعية العراقية -2019 :50: 2019 - 1301 على المقاييس المختبرية ودراسة تأثيره التآزري مع انتاج الرامنولبد بوساطة PSEUDOMONAS AERUGINOSA A3 على المقاييس المختبرية ودراسة تأثيره التآزري مع مضادات حيوية معينة مضادة لبعض البكتريا الممرضة أيوب مرتضى الشيخ فخري ناظم حسن حيدر عبد الكريم جاسم هاشم باحث أستاذ أستاذ

#### المستخلص

تهدف الدراسة إلى الحصول على كميات كافية من مستحلب الرامنولبد المنتج من عزلة محلية لبكتريا Eseudomonas aeruginosa على المستوى الصناعي، حيث استخدم المخمر الصناعي بحجم 7 لتر مع تثبيت كل الظروف المناسبة للانتاج والتي اجريت باستخدام مزارع الوجبة الواحدة. استمر الانتاج في المخمر بحجم نهائي للوسط 3.5 لتر لمدة 120 ساعة للحصول على كمية للمستحلب مساوية إلى 16 غم/لتر بدرجة حرارة 30 مئوية ودالة حموضة 6.5. استخدم راشح الوسط 120 لسراع في خفض الشد السلحي للماء من 72 الى 16 عم/لتر بدرجة حرارة 30 مئوية ودالة حموضة 6.5. استخدم راشح الوسط الانتاجي في خفض الشد السطحي للماء من 72 الى 16 غم/لتر بدرجة حرارة 30 مئوية ودالة حموضة 6.5. استخدم راشح الوسط الانتاجي في خفض الشد السطحي للماء من 72 الى 16 عمراتر وبمعامل استحلاب 76%. استخلص المنتج بالمذيبات ونقي بالسيلكا جيل باستخدام كروامتوغرافيا الزجاج ويعمود الفصل ذو الابعاد (3.5 × 30سم)، كذلك تم توصيف المنتوج المنقي بوساطة كروماتوغرافيا الطبقة الرقيقة وجهاز قياس طيف الاشعة المفصل ذو العمر 3.5% من المنتوج المنقي الرامنوليد. استخدم احماح كروامتوغرافيا الزجاج ويعمود الفصل ذو الابعاد (3.5 × 30سم)، كذلك تم توصيف المنتوج المنقي بوساطة كروماتوغرافيا الطبقة الرقيقة وجهاز قياس طيف الاشعة المفصل ذو الفصل نه 3.5% من احادي وبتنائي الرامنوليد. استخدم احادي الرامنوليد كعامل مضاد للبكتريا ضد عدد من العزلات الفصل ذو الابعاد (3.5 × 30سم)، كذلك تم توصيف المنوليد. استخدم احادي الرامنوليد كعامل مضاد للبكتريا ضد عدد من العزلات الفصل ذو الابعاد (3.5 × 30سم)، كذلك تم توصيف المنوليد. استخدم احادي الرامنوليد كعامل مضاد للبكتريا ضد عدد من الغزلات الفصل ذو الابعاد (3.5 × 30سم)، كذلك تم توصيف المنوليد. استخدم احادي الرامنوليد كعامل مضاد للبكتريا ضد عدن الغزلات تحت الحمراء حيث تبين ان المنتج مى منا حادي وبتنائي الرامنوليد. استخدم احادي الرامنوليد كعامل مضاد للبكتريا ضد عدد من الغزلات المحراء حيث تبين البرذ هو 3.5%. درس التأثري الغرض تحديد التركيز المثبط الادني، كذلك كان التأثير السمي لنفس التركيز ضد الخلايا الليفية لجنين الجرذ هو 3.5%. درس التأثير التأتري النتج من دمج احادي الرمنوليد مع مضاد الامبسلين وكان واضحا حد كل من . *وامحا هد كامن يا مديني البرذ هو 3.5%. درس التأثير الأزري الناتي من دمج* احادي الرمنوي المرفيم

الكلمات المفتاحية: مستحلب، مخمر، استحلاب، كروماتوغرافي

\*Received:18/1/2019, Accepted:8/4/2019

#### INTRODUCTION

There are multiple features which make (microbially-produced biosurfactants alternative for surfactants) a promising chemically-synthesized ones in industrial applications, owing these features to their higher effectiveness and biodegradability at pН temperature. extreme and salinity conditions, lower toxicity and environmental friendly. These molecules are biosynthesized by a number of microorganisms (21). Pseudomonas aeruginosa seems to produce rhamnolipid to some degree from substrates regardless of their hydrophobicity. This microorganism adapted for growing and biosynthesize surfactants in complex forms such as wastes and soil systems (22). The cellular metabolic pathway is directed toward rhamnolipids production at the end of exponential growth phase when nitrogen has been completely consumed (28). There's a big optimization of fermentation need for conditions in order to increase the product vield of biosurfactants as well as improving their specificity to overcome the barriers that limit industrial production and exploiting of microbial surfactants (26). The most abundant forms of rhamnolipids are rhamnosyl-Bhydroxydcanoyl-B-hydroxydecanoate (Rha- $C_{10}$ - $C_{10}$ ), which is a mono-rhamnolipoid, and rhamnosyl-rhamnosyl-B-hydroxydcanoyl-Bhydroxydecanoate (Rha-Rha-C<sub>10</sub>-C<sub>10</sub>), a dirhamnolipid. Although, some rhamnolipids contain one or two rhamnose moiety linked to only one 3-hydroxyfatty acid (37). The activity of RL depends on the surface active compound quantity until the concentration of critical micelle (CMC) is obtained. Based on the theories of classic surfactant aggregation, the molecules of surfactant are integrate into the structure of water, which called monomers in aqueous phase, and aggregate at the interface of air-liquid or liquid-liquid at concentrations less than CMC (14). Once RL molecules reach its threshold, they associate to each other to form aggregates. These micelles when form in water, their tails, which are hydrophobic, form a core that encapsulates the hydrocarbons, and the hydrophilic polar heads maintain their contact with water by forming an outer shell. In the case of oil, the process is reversed (reversed micelle) where the hydrophobic part forms the outer shell and the polar head inside (29). Furthermore, various structures autoassembled by RL when the concentration above the CMC, such as vesicles or lamella, depending on the concentration of RL and pH value (19). Three sequential reactions involved in the production of rhamnolipids. The first, which is responsible of synthesis of fatty acid dimer moiety, carried out by an enzyme, RhlA. The second reaction is catalyzed by RhlB, a membrane-bound rhamnosyltransferase, uses 3-(hydroxyalkanoyloxy) alkanoic acid (HAA) and deoxythymidine diphospho-l-rhamnose (dTDP-L-rhamnose) as precursors, producing mono-rhamnolipids. These are in turn the substrates, jointly with dTDP-L-rhamnose, producing di-rhamnolipids by the loosely bound to inner membrane, RhlC (27, 37). Glycolipids conquer a minor part of the molecular structure of the cell and commonly placed with lipids associated with external cell surfaces. Interestingly, glycolipids assist in molecular transfer across cell membrane as well as known to be involved in the biosynthesis proteoglycans of and glycoproteins (12). There are an increasing number of pathogens which are unsusceptible to treatment with common antibiotics; this challenges human health as well as other organisms. Therefore, the need for searching and development of novel pharmaceuticals characterized by a wide spectrum of activity, low toxicity as well as cost effective that can be applied for human, veterinary medicine and plant is of great importance (16).Biosurfactants can disturb the structure of biological membranes due to their interaction with membranous proteins as well as phospholipids (43). One of the major problems of hydrophobic antibiotics is that the difficulty in their uptake bu cells. So RL are tested for enhancing the activity of such hydrophobic antimicrobials (25). The aim of the present work is to produce rhamnolipid from P. aeruginosa A3 in bioreactor and testing its synergistic effect with antibiotics against resistant clinical isolates.

#### MATERIALS AND METHODS

#### **Production of RL in Bioreactor**

The bench-top LAMBDA MINIFOR laboratory bioreactor was used for RL production in optimized conditions. This

model is easy to handle and all important cultural conditions can be measured and controlled. The minimum working volume was 3.5 L of the 7 L capacity of the bioreactor vessel. The instrument is equipped with different sensitive probes for temperature, pH, air and agitation system The production medium was composed from the modified Proteose-peptone ammonium salts (PPAS), as follows (g/l): NH<sub>4</sub>Cl, 1.0; KCl, 1.5; MgSO<sub>4</sub>, 0.19; Tris HCl, 14.5; proteose-peptone 2.0; CTAB, 0.2; the media was autoclaved for 15 mis at 121°C after the addition of 2.5% glycerol as carbon source, then 0.1% (v/v) trace elements solution composed of (g/l): FeSO4.7H<sub>2</sub>O, 2.0; MnSO<sub>4</sub>.H2O, 1.5: (NH<sub>4</sub>)6Mo7O24.4H<sub>2</sub>O, 0.6; ZnSO4.7H<sub>2</sub>O, 0.29; and CuSO4.5H<sub>2</sub>O, 0.25; sterilized by cellulose acetate Millipore filter (0.22µm), was added to the medium when warmed down after autoclaving (35). Seventy ml of the previous production medium (2%) that contain  $1 \times 10^6$ cfu/ml was used to inoculate the bioreactor containing same production medium with the daily addition of 5 ml of the remaining spent medium after sterilization with cellulose acetate Millipore filter (0.22µm) as a good elicitor for RL production.

#### **Biosurfactant extraction**

The broth culture was extracted from biomass by centrifugation 10,000 rpm for 15 mins, then the pH of crude culture was adjusted by 1 N HCl to pH of 2, left overnight for precipitation, the latter was collected through centrifugation at 10,000 rpm for 20 mins, while the above supernatant extracted with an equivalent volume of chloroform: methanol (2:1 v/v). The median scum layer was separated and the superior aqueous phase re-extracted with similar volumes of solvents above. The merged extracts were concentrated by evaporation of solvents at 45° C to obtain viscous honey-colored extract. The yields were determined gravimetrically in terms of g/L (36, 41).

#### **Biosurfactant purification**

The RLs components were separated from the crude mixture as follows. Heat activated silica gel 60 slurry in chloroform was decanted onto a column of a glass chromatography ( $3.5 \times 30$ cm). Two grams of crude extracted RL were dissolved in chloroform (4 ml) and added onto

the column. Then the column washed with chloroform at a flow rate of 60 ml/hr to elute neutral lipids totally, followed by two steps of chloroform/methanol addition at volumes of 250 and 200 ml with 50:3 and 50:5 ratios, respectively, for separation of mono-RL. Final elution was done by same solvents above at 50:50 ratios with pure methanol, each volume of 100 ml, to elute di-RL. The fractions composition was checked by thin layer chromatography (34).

#### Analyzing of biosurfactant

#### Thin Layer Chromatography (TLC)

Thin layer chromatography (TLC) technique was used to characterize the biosurfactants purification to determine their types. The process was conducted by TLC plates ( $10 \times 10$ cm) coated with silica gel. Separation and analysis of the produced biosurfactant proceed through а solvent system comprised chloroform, methanol and acetic acid in a ratio of 65:15:2, respectively. Then the plates were sprayed with resolving solution composing of 0.15g of orcinol dissolved in 8 ml of 60% H<sub>2</sub>SO<sub>4</sub> and the final volume completed to 50 ml with deionized distilled water, left for 10 mins in oven at 110°C. The appearance of brown spots was considered as positive test for the glycolipid biosurfactant. The spots were compared with standard RL and according to the R<sub>f</sub> value which is rate of retention flow calculated by the formula:

 $R_f$ = Distance of sample spot/ Distance of solvent (42).

#### Fourier Transform Infrared Spectroscopy (FTIR) spectral analysis

The functional groups and chemical bonds were illustrated using FTIR spectrometry (Shimadzu 8400, Japan). The spectrum was limited at the range of 4000-650 cm<sup>-1</sup> with resolution of 4 cm<sup>-1</sup>

#### Surface tension measurement

The surface tension of an aqueous solution was measured by the Wilhelmy platinum plate with a QBZY-2 Tensiometer (China). Fifteen ml of supernatant was poured into 50 ml glass beaker and put onto the tensiometer platform. The measurement was carried out at  $25\pm1$  C after dipping the plate in the solution until monitoring the value of supernatant surface tension following the procedure of measurement written in the manual of the instrument. Between each measurement, the Wilhelmy plate was rinsed with acetone and burned by alcohol burner. For more accurate value, the average of three records was used in the study (32).

## Critical Micelle Concentration (CMC) and critical micelle dilution (CMD)

For CMC and CMD analysis, produced RL were subjected to serial twofold dilutions then measuring the surface tension. CMC is the minimum concentration that reduces the surface tension to the lowest level, while CMD is the maximum dilution that reduces the surface tension to the bottom level (7). The CMD was determined by ten-fold dilution, CMD<sup>-1</sup> and CMD<sup>-2</sup>. For measuring critical micelle concentration (CMC) of rhamnolipid after purification, concentrations range (5-500 mg/L) was prepared at room temperature (23). For a positive control, the CMC of sodium dodecyl sulfate (SDS) (100-10,000 mg/L) was prepared while the negative control is the distilled water. The CMC as well as the surface tension at the point of CMC were specified from the cut-point in the surface tension of these solutions and recorded (31).

#### Dry weight cell determination

After cultivation for 5 days, the fermentation broth was centrifuged at 8,500 g for 20 min to remove bacterial cells. 10 mL of the supernatant was taken to detect emulsification index ( $E_{24}$ ) and surface tension. The collected bacterial cells allowed to dry in oven to obtain the dry weight, which is reported in terms of g/L (6).

# Measurement of antibacterial activity of mono-rhamnolipid

Antimicrobial activity of purified mono-RL determined according to values of was minimum inhibitory concentration (MIC), which is the lowest concentration of biosurfactant needed to inhibit the visible growth of a given organism on Mueller Hinton agar after incubation time. A method of twofold serial dilution of rhamnolipid was used to quantify antimicrobial activity. The process was assessed against S. aureus, Klebsiella pneumoniae, Proteus mirabilis, Streptococcus iniae, E. coli, Serratia marcescens and P. aeruginosa. The susceptibilities were specified by micro-dilution technique using 96 well plates (U-shaped). A volume of 200ul of stock

for each solution  $(\mu g/ml)$ rhamnolipid dissolved in Mueller-Hinton broth (MHB) was transferred into the first well of each row and two-fold serially diluted in subsequent wells by mixing with 100ul of MHB. A bacterial suspension of 100ul  $(1 \times 10^6 \text{ cfu/mL})$  was added to these wells to extent final inoculum size of  $5 \times 10^5$  cfu/well and incubated for 24 h at 37°C. The plates tested for visible growth and compared with rhamnolipid-free control wells. The MIC determined as the lowest concentration which the at tested microorganism does not pretend visible growth (33). All tests were conducted in triplicate.

#### Maintenance of cell cultures

Rat embryonic fibroblast (REF) Cell lines were maintained in RPMI-1640 medium supplemented with 10% Fetal bovine, penicillin 100 µg /mL, and streptomycin 100 µg/mL. Cells were passage using Trypsin-EDTA reseeded at 80% confluence twice a week, and incubated at 37 °C to perform cytotoxicity assay for produced the rhamnolipid (40).

### Cytotoxicity Assays

For determining the cytotoxic effect of mono-RL, the assay of MTT cell viability was done with the aid of 96-well plates. Cell lines were seeded at  $1 \times 10^4$  cells/well. After 24 h. or achieving confluent monolayer, cells were treated with RL. The viability of cells was measured after 72 h of treatment by discarding the medium, adding 28 µL of 2 mg/mL solution of MTT and then reincubating the cells for 2.5 h at 37 °C. After removing the MTT solution, the crystals remained in wells were solubilized through the addition of 130 µL of DMSO Sulphoxide) (Dimethyl followed bv incubation for at 37 °C for 15 min in oscillatory shaker (1). The absorbency at 492 nm (test wavelength) was determined on a microplate reader; the assay was conducted in thrice. The rate of cell growth inhibition percentage of cytotoxicity) (the was calculated as the following equation:-

#### Cytotoxicity = A-B/A \*100

Where A and B are the optical density of control and test, respectively

#### Synergistic effect with antibiotics

The purified mono-RL was tested for synergistic effect with some antibiotics against selected pathogenic isolates in well-plate method after combination, such as ampicillin, cefotaxime, gentamycin, azithromycin, tetracycline, vancomycin. Ciprofloxacin was used as positive test for antibiotic activity against same isolates. Both MIC and CMC of RL were mixed with same concentration of antibiotic discs used in this experiment and placed in pre-cut wells of Muller Hinton agar cultured with pathogenic agents.

#### **RESULTS AND DISCUSSION**

#### **Production of RL in bioreactor**

After 120 h of incubation at 30°C with steady favorite conditions, the surface tension of solution reached its minimum value (27.2 mN/m) with emulsification activity 67% and biomass 2.7 g/l. The RL concentration was 16 g/l (Figure 1).

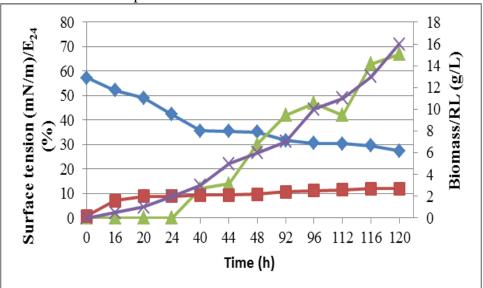


Figure 1. Production of RL by *P. aeruginosa* A3 in bioreactor; biomass(■); RL, (×); surface tension, (�); E24%, (▲)

The dry cell weight was nearly constant after 24 hrs of production till the end of the experiment, while there was a continual increasing in RL production as well as emulsification index, whereas the surface tension decreased gradually. This experiment confirms that RL is a secondary metabolite. The maximum RL concentration produced by *P. aeruginosa* in a lab-scale fermenter was 22.5 g/L with biomass concentration was 5.3 g/L (3), while Noh *et al.*, (24) obtained 23.6 g/L of RL from *P. aeruginosa* USM-AR2.

## Analysis of biosurfactant by thin layer chromatography

After purification of crude RL by column chromatography, the separated fractions were analyzed using TLC. The spots appeared on the TLC plates were corresponding to rhamnsoe domains (R<sub>f</sub> value 0.41 and 0.17) which represent mono- and di-rhamnolipids respectively (Figure 2). This indicates that this isolate (P. aeruginosa A3) produce a mixture of mono- and di-RL. The preliminary characterization of this type of biosurfactant was a glycolipid (13). Three distinct peaks (0.17, 0.34 and 0.53) were observed in the study conducted by Bharali et al., (5) to produce RL from P. aeruginosa grown on biodiesel, while only two spots (Rf of 0.9 and 0.6) observed by Das et al., (10) for mono and di-RL, respectively. These differences in R<sub>f</sub> values is correlated to the congeners of the biosurfactant mixture depending on the production isolate and growth conditions.

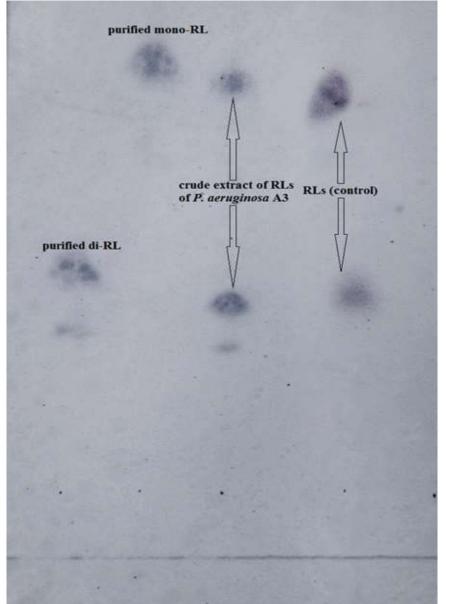
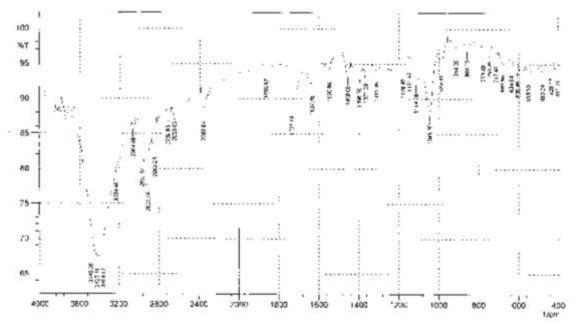


Figure 2. Analyzing of rhamnolipid by TLC

Fourier Transform Infrared Spectroscopy (FTIR) spectral analysis

Many major peaks are appeared in the spectrum of mono-RL, which are the (C-H) model between 2825 and 2954 cm-1, and the carboxylic acids extending between 1500 and 1740 cm-1. The broad peak at 1737 cm-1 comprises both the carboxylic acid part. At 1560 cm-1, the band corresponds to the COO of the rhamnolipid. The region below 1500 cm-1

contains overlapping peaks assigned to specific frequencies of RL group. Peaks at 1461 and 1379 cm-1 related to the (CH) modes, and the bands series between 1100 and 1300 cm-1 are associated with ester vibrations (C-O-C) (Figure 3). The total mono-RL concentration can be related to the absorbency at 1738 for the ester (C=O) band (17). The C-H and O-H groups refer to the rhamnose molecules of RL (18).



**Figure 3. Characterization of RLs by FTIR, A: mono-RL, B: di-RL Critical Micelle Concentration (CMC)** It was obtained that 250 µg/mL

The mono-RL produced by *P. aeruginosa* A3 was able to reduce the surface tension of distilled water to 27 mN/m when the CMC value is 40 mg/L. This means that only 40mg/L of mono-RL is needed to reduce the surface tension to the minimum level. These values were different with Samadi *et al.*, (33) who found that 20 mg of RL was able to reduce the surface tension to 24mN/m. However; the synthetic surfactants generally had a higher CMC values that RL. Sodium dodecyl sulfate show CMC of 2200 mg/L. The results have shown the effectiveness of mono-RL produced by *P. aeruginosa* A3 isolate. **MIC** 

It was obtained that 250  $\mu$ g/mL was the MIC for most of bacterial used in the experiment; however, there's no MBC effect for the rhamnolipid even when the concentration was 1 g/L. This confirms that no bactericidal effect for the RL (Figure 4). These results are consistent with the cytotoxicity of RL on REF applicable cell line. where the toxic concentration was 23% corresponding to 250 µg/ml. The MIC of Mono-RL produced from P. aeruginosa MN1 against several Grampositive bacteria was 25 mg/L, while it needed more than 1600 mg/L to affect the Gramnegative bacteria. However, RL-mixture and di-RL needed 50 mg/L to inhibit the Grampositive bacteria (33).



Figure 4. The effect of purified mono-RL on growth of bacterial cells, A, S. aureus; B, E. coli; C, Serratia marcescens; D, Streptococcus iniae; E, Klebsiella pneumoniae; F, Proteus mirabilis; G, P. aeruginosa, H, control.

#### Cytotoxicity effect

The cytotoxicity of mono-RL in concentration of 250  $\mu$ g/mL (the same concentration used for MIC experiment) was 23% on REF cell line (Figure 5), this concentration is considered as acceptable for animal cell line because of low toxicity which gives a strong indication that mono-RL could be used as antimicrobial therapy. Also, the higher concentrations (500 and 1000  $\mu$ g/mL) are less than 50% toxicity, which gives an indication about the high difference between the toxic dose and effective dose for antimicrobial therapy.

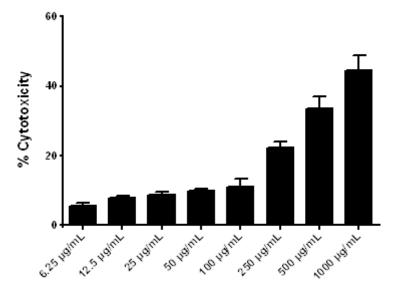


Figure 5. Cytotoxicity effect of mono-RL in REF cell line

# Synergistic effect of mono-RL with antibiotics

The requirement for new antimicrobial agents to beat bacterial antibiotic-resistance leads to investigation for modern antimicrobial strategies. Thus, the combination of different antimicrobial compounds in order to increase their efficacy is considered as a magnificent solution for the problem. A synergistic association is when two or more agents act to enhance efficacy compared with single ones (11). Probably, the biosurfactant interfere with bacterial membranes and forming pores that permit the entrance of antibiotics into the cells, which increase the antimicrobial agent activity, owing to the amphiphilic nature of the surfactants, which destabilize lipid packing of biological membranes, altering integrity and penetrate these coats through hydrophobic interactions, that lead ultimately to an increase of the antibiotic impact (15). The best combination effect was found when clinical isolates were resistant to both mono-RL and ampicillin, but sensitive when these two antimicrobial agents were combined together

in their MIC (Table 1 and Figure 6) (8, 9). A similar study conducted to demonstrate that concentrations of RL lower than CMC could increase the quantity of anionic phospholipids in membranes of B. subtilis 168 that lead to more sensitivity to sulfonates antimicrobials (39). The major differences in the composition of phospholipids in different bacterial species, and this will ultimately determine their response to antimicrobial compounds. A research conducted by Sotirova et al., (38) was demonstrated that RL in concentration beneath the CMC could decrease the number of outer mambrane proteins (OMPs) of P. aeruginosa NBIMCC 1390 that lead to changes in pathogenisity and antibiotic sensitivity. These effects include the proteins composition and ultimately alter the membranous organization pathogens under investigation. of Anv reduction in the OMPs could decrease the tightness of the outer membrane and, finally, increasing the passage of hydrophobic compounds, but still has no effect on cell viability, unless accompanied with another bactericidal agent.

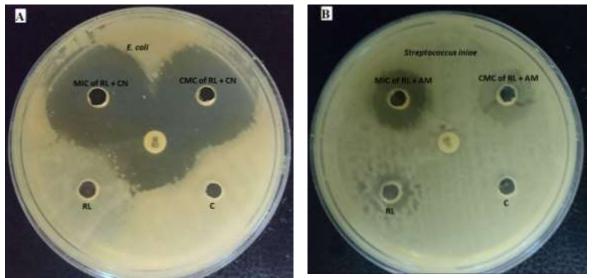


Figure 6. Combination effect of mono-RL with antibiotics against clinical isolates A: *E. coli*, B: *Streptococcus iniae*, CN: gentamycin, AM: ampicilin, C: control Table1. Synergistic effect of mono-RL with different antibiotics

Ampicillin disc (25µg/ml)			
	(250µg/ml)	(250µg/ml)	(40µg/ml)
S	R	S	S
R	R	S	S
R	R	S	S
R	R	S	Ι
S	R	S	S
S	S	S	S
I	I	S	S
Cefotaxime disc (30µg/ml)	Mono-RL	MIC combine	CMC of RL combine
I	R	S	S
R	R	S	S
R	R	S	R
S	R	S	S
R	R	S	S
S	S	S	S
R	I	S	S
Gentamycin disc (30µg/ml)	Mono-RL	MIC combine	CMC of RL combine
R	R	S	S
S	R	S	S
S	R	S	S
Ι	R	S	Ι
S	R		S
S	S		S
ŝ	Ĩ		ŝ
Azithromycin disc (15µg/ml)	Mono-RL		CMC of RL combine
S			S
S	R		S
	R		S
Ĩ	R		Ĩ
R			s
			š
~			š
~	=		CMC of RL combine
R	R		I
R	R		S
			Ĩ
I			Ī
Ī			Ĩ
			ŝ
I			Š
-	Mono-RL	MIC combine	CMC of RL combine
S	R	S	S
Ř	R		ŝ
R	R	ŝ	ŝ
			Ĩ
			Î
			ŝ
R	Ĩ	š	Š
	R R R S S S I Cefotaxime disc (30µg/ml) I R R S S R Gentamycin disc (30µg/ml) R S S S S Azithromycin disc (15µg/ml) S S S S I R R S S S Teteracyclin disc (30µg/ml) R R R I I R S S S S Teteracyclin disc (30µg/ml) R R R R R R R R R R R R R R R R R R R	R R R R R R R R R R S S I I I Cefotaxime disc (30µg/ml) Mono-RL I R R R R R R R S S R I Gentamycin disc (30µg/ml) Mono-RL R R S R S R I R S R S R S R I R S R S R S R S R S R S R S R S	RRSRRSSRSSRSSSSIISCefotaxime disc (30µg/ml)Mono-RLMIC combineIRSRRSRRSSRSSRSSRSSRSSRSGentamycin disc (30µg/ml)Mono-RLMIC combineRSSSSRSSRSSRSSRSSRSSRSSRSSRSSRSSRSSRSSS </td

Lipopolysaccharide has an important role in the resistance of bacteria to antimicrobial agents (30). The type and amount of LPS is a crucial factor in the interactions between bacteria and their environment (20). Any loss in LPS due to exposure to RL, even at low concentrations, resulted in increasing of cell surface hydrophobicity (2). The liberation of form cell surface happened after LPS solubilization of the outer membrane due to binding and aggregating of biosurfactant on membrane (38). Applying RL the in concentration much below its CMC could affect the composition of bacterial membranes by increasing their permeability, and, resulting in bactericidal effects of antibiotics at lower concentrations than the accepted worldwide (39). In this study, the optimum conditions for RL production by local Pseudomonas aeruginosa isolate was conducted in a bioreactor for industrial-scale production to obtain 16 g/L of biosurfactant. Extraction and purification carried out to conclude that this RL is consists of both mono- and di-RL. The antibacterial activity and cytotoxicity of mono-RL was performed to assign specific concentration for further experiments, which µg/ml; this concentration was was 250 combined with some antibiotics to overcome the resistance of some clinical isolates to such antibacterial agents.

### REFERNCES

1. Al-Shammari A.M., M.I. Salman, Y.D. Saihood, N.Y. Yaseen, K. Raed, H.K. Shaker, A. Ahmed, A. Khalid and A. Duiach. 2016. In vitro synergistic enhancement of Newcastle disease virus to 5-fluorouracil cytotoxicity against tumor cells. Biomedicines. 4(1): 3

2. Al-Tahhan, R.A., T.R. Sandrin, A.A. Bodour and R.M. Maier. 2000. Rhamnolipidinduced removal of lipopolysaccharide from *Pseudomonas aeruginosa*: effect of cell surface properties and interaction with hydrophobic substrates. Applied Environmental Microbiology. 66: 3262-3268.

3. Bazsefidpar, S., B. Mokhtarani, R. Panahi and H. Hajfarajollah. 2019. Overproduction of rhamnolipid by fed-batch cultivation of *Pseudomonas aeruginosa* in a lab-scale fermenter under tight DO control. Biodegradation. 30(1): 59-69 4. Benincasa, M., A. Marqués, A. Pinazo and A. Manresa. 2010. Rhamnolipids surfactants: alternatives substrates, new strategies. In. Sen, R. (ed.). *Biosurfactants*, pages. 170 - 184. New York: Landes Bioscience and Springer Science 5. Bharali, P., S.P. Singh, N. Dutta, S. Gogoi, L.C. Bora, P. Debnath and B.K. Konwar. 2014. Biodiesel derived waste glycerol as an economic substrate for biosurfactant production using indigenous *Pseudomonas aeruginosa*. RSC Advances. 4: 38689-38706

6. Bordoloi, N.K. B.K. Konwar. 2008. Microbial surfactant-enhanced mineral oil recovery under laboratory conditions. Colloids and Surfaces B: Biointerfaces. 63: 73-82

7. Cheng, T., J. Liang, J. He, X. Hu, Z. Ge and J. Liu. 2017. A novel rhamnolipid-producing *Pseudomonas aeruginosa* ZS1 isolate derived from petroleum sludge suitable for bioremediation. AMB Express. 7(120): 1-14.

8. CLSI. 2012. Methods for dilution antimicrobial susceptibility tests for bacteria that grow aerobically. 9th ed. CLSI standard M07-A9. Wayne, PA: Clinical and Laboratory Standards Institute.

9. CLSI. 2018. Performance Standards for Antimicrobial Disk Susceptibility Tests. 13th ed. CLSI standard M02. Wayne, PA: Clinical and Laboratory Standards Institute.

10. Das, P., X. Yang and L.Z. Ma. 2014. Analysis of biosurfactants from industrially viable Pseudomonas strain isolated from crude oil suggests how rhamnolipids congeners affect emulsification property and antimicrobial activity. Frontiers in Microbiology. 5(696): 1-8.

11. Fair, R.J. and Y. Tor. 2014. Antibiotic and bacterial resistance in the 21<sup>st</sup> century. Perspectives in Medicinal Chemistry. 6: 25-64. 12. Fowler, Z.L., E. Leonard and M. Koffas. 2010. Biosynthesis of cellular building blocks: The prerequisites of life. In: Smolke, D.S. (ed.). The metabolic pathway engineering handbook, Fundamentals. London. CRC press. pp: 21

13. George, S. and K. Jayachandran. 2012. Production and characterization of rhamnolipid biosurfactant from waste frying coconut oil using a novel *Pseudomonas aeruginosa* D. Journal of Applied Microbiology. 114: 373-383. 14. Guo, Q., J. Yan, J. Wen, Y. Hu, Y. Chen and W. Wu. 2016. Rhamnolipid-enhanced aerobic biodegradation of triclosan (TCS) by indigenous microorganisms in water-sediment systems. Science of the Total Environment. 571: 1304-1311

15. Joshi-Navare, K. and A. Prabhune. 2013.
A biosurfactant-sophorolipid acts in synergy with antibiotics to enhance their efficiency.
BioMed Research International.
2013(512495): 1-8

16. Lazarkevich, I., A. Sotirova, T., Avramova and D. Galabova. 2015. Comparative study on antibacterial activity of synthetic analogues of biologically active compounds and their combination with rhamnolipid biosurfactant. Journal of Bioscience and Biotechnology. SE/ONLINE: 55-62

17. Lebron-Paler, A. and J.E. Pemberton. 2006. Determination of the acid dissociation contstant of the biosurfactant monorhamnolipid in aqueous solution by potentiometric and spectroscopic methods. Analytical Chemistry. 78: 7649-7658.

18. Leitermann, F., C. Syldatk and R. Hausamnn. 2008. Fast quantitative determination of microbial rhamnolipids from cultivation broths by ATR-FTIR spectroscopy. Journal of Biological Engineering. 2(13): 1-8

19. Ma, Y., Y. Li, C. Huang, Y. Tian and Z. Hao. 2018. Rhamnolipid biosurfactants: functional properties and potential contributors for bioremediation. Biodegradation. 30(1): 1-11

20. Makin, S.A. and T.J. Beveridge. 1996. The influence of A-band and B-band lipopolysaccharide on the surface characterisitics and adhesion of *Pseudomonas aeruginosa* to surfaces. Microbiology. 142: 299-307.

21. Marchant, R. and I.M. Banat. 2012.Biosurfactants: A sustainable replacement for chemical surfactants? Biotechnology Letters. 34(9): 1597-1605

22. Mata-Sandoval, J.C., J. Karns and A. Torrenrs. 1999. High-performance liquid chromatography method for the characterization of rhamnolipid mixtures produced by *Pseudomonas aeruginosa* UG2 on corn oil. Journal of Chromatography. A. 864: 211-220

23. Morikawa, M., Y. Hirata and T. Imanaka. 2000. A study on the structure and function relationship of lipopeptide biosurfactants. Biochimica et Biophysica Acta (BBA) -Molecular and Cell Biology of Lipids. 1488: 211-218.

24. Noh, N.A.M., S.M. Salleh and A.R.M. Yahya. 2013. Enhanced rhamnolipid production by *Pseudomonas aeruginosa* USM-AR2 via fed-batch cultivation based on maximum substrate uptake rate. Letters in Applied Microbiology. 58: 617-623

25. Noordman, W.H. and D.B. Janssen. 2002. Rhamnolipid stimulates uptake of hydrophobic compunds by *Pseudomonas aeruginosa*. Applied Environmental Microbiology. 68(9): 4502-4508

26. Perfumo, A., I.M. Banat, F. Canganella R. Marchant. 2006. Rhamnolipid and thermophilic production by а novel hydrocarbon degrading Pseudomonas aeruginosa APO2-1. Applied Microbiology and Biotechnology. 72: 132-138.

27. Rahim, R., U.A. Ochsner, C. Olvera, M. Graninger, P. Mesner, J.S. Lam and G. Sorbeon-Chavez. 2001. Cloning and functional characterization of the Pseudomonas aeruginosa rhlC gene that encodes rhamnosyltrans ferase 2, an enzyme responsible for di-rhamnolipid biosynthesis. Molecular Microbiology. 40(3): 708-718

28. Rashedi, H., E. Jamshidi, M.M. Assadi and B. Bonakdarpour. 2005. Isolation and production of biosurfactant from *Pseudomonas aeruginosa* isolated from Iranian southern wells oil. International Journal of Environmental and Science Technology. 2(2): 121-127.

29. Rodrigues, R.L. 2015. Microbial surfactants: fundamentals and applicability in the formulation of nano-sized drug delivery vectors. Journal of Colloids and Interface Science. 449: 304-316.

30. Russell, A.D. 1995. Mechanisms of bacterial resistance to biocides. International Biodeterioration and Biodegradation. 36: 247-265

31. Sabturani, N., J. Latif, S. Radiman and A. Hamzah. 2016. Spectroscopic analysis of rhamnolipid produced by *Pseudomonas aeruginosa* UKMP14T. Malaysian Journal of Analytical Sciences. 20(1): 31-43 32. Saikia, R.R., S. Deka, M. Deka and H. Sarma. 2012. Optimization of environmental factors for improved production of rhamnolipid biosurfactant by *Pseudomonas aeruginosa* RS29 on glycerol. Journal of Basic Microbiology. 52: 446-457

33. Samadi, N., N. Abadian, R. Ahmadkhaniha, F. Amini, D. Dalili, N. Rastkari, E. Safaripour and F.A. Mohseni. 2012. Structural characterization and surface activities of biogenic rhamnolipid surfactant from Pseudomonas aeruginosa isolate MN1 and synergistic effects against methicillin-Staphylococcus resistant aureus. Folia Microbiologica. 57(6): 501-508

34. Sanchez, M., J.A. Teruel, M.J. Espuny, A. Marques, F.J. Aranda, A. Manresa and A. Ortiz. 2006. Modulation of the physcial properties of

dielaidolphosphatidylethanolamine

membranes by a dirhamnolipid biosurfactant produced by *Pseudomonas aeruginosa*. Chemistry and Physics of Lipids. 142: 118-127.

35. Siegmund, I. and F. Wagner. 1991. New method for detecting rhamnolipids excreted by *Pseudomonas* species during growth on mineral agar. Biotechnology Techniques. 5: 265-268

36. Sim, L., O.P. Ward and Z.Y. Li. 1997. and characterization of Production a biosurfactant isolated from Pseudomonas aeruginosa UW-1. Journal of Industrial Microbiology and Biotechnology. 19: 232-238 37. Sorbeon-Chavez, G., F. Lepine and E. Deziel. 2005. Production of rhamnolipids by aeruginosa. Pseudomonas Applied Microbiology and Biotechnology. 68: 718-725 38. Sotirova, A., D. Spasova, E. VasilevaTonkova and D. Galabova. 2009. Effects of rhamnolipid-biosurfactant on cell surface of *Pseudomonas aeruginosa*. Microbiological Research. 164: 297-303

39. Sotirova, A., T. Avramova, S. Stoitsova, I. Lazarkevich, V. Lubenets, E. Karpenko and D. Galabova. 2012. The importance of rhamnolipid-biosurfactant-induced changes in bacterial membrane lipids of *Bacillus subtilis* for the antimicrobial activity of thiosulfonates. Current Microbiology. 65(5): 534-541.

40. Sulaiman, G.M., M.S. Jabir and A.H. Hameed. 2018. Nanoscale modification of chrysin for improved of therapeutic efficiency and cytotoxicity. Arificial Cells, Nanomedicine, and Biotechnology. 46(sup1): 708-720.

41. Thaniyavaran, J., A. Chongchin, N. Wanitsuksombut, S. Thaniyavaran, P. Pinphanichakaran, N. Leepipatpiboon, M. Morikawa and S. Kanaya. 2006. Biosurfactant production by *Pseudomonas aeruginosa* A41 using palm oil as carbon source. The Journal of General and Applied Microbiology. 52:215-222.

42. Wittgens, A., T. Tiso, T.T. Arndt, P. Wenk, J. Hemmerich, C. Muller, R. Whichmann, B. Kupper, M. Zwick, S. Wilhelm, R. Hausamnn, C. Syldatk, F. Rosenau and L.M. Blank. 2011. Growth independent rhamnolipid production from glucose using the non-pathogenic Pseudomonas putida KT2440. Microbial Cell Factories. 10(80): 1-17

43. Yilmaz, E.S. and U. Sidal. 2005. Investigation of antimicrobial effects of a *Pseudomonas*-originated biosurfactant. Biologia. 60(6): 723-725.